

Answer all questions in the spaces provided. On numerical problems, *show your work* (set up the problems dimensionally). Be careful to express answers to the *appropriate number of significant figures* and include units where appropriate. Point values are in parentheses.

1.(10) Identify the following as either elements (E), compounds (C) or mixtures (M):

- a) neon E
- b) blood M
- c)  $C_{12}H_{22}O_{11}$  (sugar) C
- d) brass M

2.(10) Determine whether the following are physical (P) or chemical (C) changes:

- a) dissolving sugar into water P
- b) conversion of water into hydrogen and oxygen C
- c) steam condensing into droplets P
- d) digestion of food C

3.(10) Using the terms “shape” and “volume”, and any combination of the modifiers “definite” and “indefinite”, define (e.g. “definite shape, definite volume”):

a gas – indefinite shape, indefinite volume

a liquid – indefinite shape, definite volume

4.(15) a) Express the number 5,523.6 to four significant figures using scientific notation.

$5.524 \times 10^3$

b) Express 0.000 000 002 105 to two significant figures using scientific notation.

$2.1 \times 10^{-9}$

c) If your answer in b) is in units of g, express it in units of mg (stay in scientific notation).

$2.1 \times 10^{-9} \text{ g} \times [10^3 \text{ mg/g}] = 2.1 \times 10^{-6} \text{ mg}$

5.(10) *Determine the density* of mercury (Hg) if 225.8 g occupies a volume of 16.6 mL. Show your work and be sure to include units.

Density = mass/volume  
= 225.8 g/16.6 mL  
= 13.6 g/mL

6.(10) A jumbo family size bottle of children's decongestant contains 1.69 pints. *What is this volume in units of mL? Show your work.*

$$1.69 \text{ pt} \times 1 \text{ qt}/2 \text{ pt} \times 0.9464 \text{ L}/1 \text{ qt} \times 1000 \text{ mL}/1 \text{ L} = \mathbf{800 \text{ mL}}$$

**or  $8.00 \times 10^2 \text{ mL}$**   
(3 sig. figures)

7.(10) A premature baby weighs 1.12 kg. *How many ounces (oz) does the infant weigh? Show your work.*

$$1.12 \text{ kg} \times 2.205 \text{ lb}/1 \text{ kg} \times 16 \text{ oz}/1 \text{ lb} = \mathbf{39.5 \text{ oz}}$$

(3 sig. figures)

#### Bonus

8.(10) Often, a lethal dose of a toxin is measured in g of toxin / kg body weight of a specific animal population. The dose that will be lethal for 50% of a population is called an LD<sub>50</sub>. The dose of ethanol (grain alcohol) that is lethal for 50% of humans (the LD<sub>50</sub>) is 1.70 g ethanol / kg body weight.

a) What is a lethal dose in g for 50% of those humans weighing 82.5 kg?

A lethal dose is 1.70 g for every kg body weight.  
 $82.5 \text{ kg} \times 1.70 \text{ g}/1 \text{ kg} = \mathbf{140 \text{ g ethanol}}$

b) The density of ethanol is 0.80 g/mL. What is the lethal dose in mL?

Density = mass/volume solve for Volume  
Volume = mass/density  
Or set it up dimensionally  
 $140 \text{ g} \times 1 \text{ mL}/0.80 \text{ g} = \mathbf{180 \text{ mL ethanol}}$   
2 sig. figures

#### Conversion factors (you will not need all of these)

Length:

$$\begin{aligned} 1 \text{ m} &= 39.37 \text{ in} \\ 1 \text{ in} &= 2.54 \text{ cm} \\ 1 \text{ mi} &= 1.609 \text{ km} \end{aligned}$$

Volume:

$$\begin{aligned} 1 \text{ qt} &= 0.9464 \text{ L} \\ 1 \text{ qt} &= 2 \text{ pt (exact)} \end{aligned}$$

Mass:

$$\begin{aligned} 1 \text{ kg} &= 2.205 \text{ lb} \\ 1 \text{ oz} &= 28.35 \text{ g} \\ 1 \text{ lb} &= 16 \text{ oz (exact)} \end{aligned}$$

Density is units of mass per unit volume