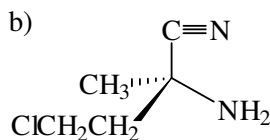
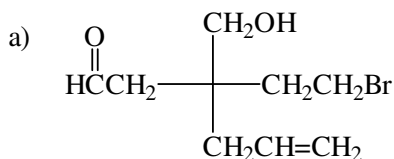
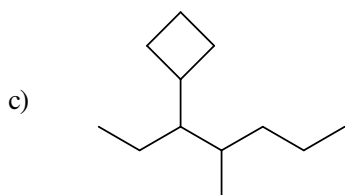
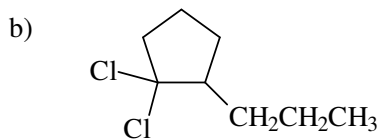
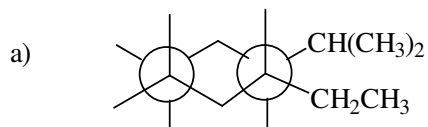


1.(12) In the following molecules, **rank the groups** in the molecules attached to the chiral centers in order of Cahn, Ingold, Prelog (CIP) priority (**1 = highest; 4 = lowest priority**) and assign the absolute configuration (**R or S**). For full credit, you must rank the groups 1-4.



2.(12) Name the following compounds. Be sure to denote stereochemistry (*cis*, *trans*) where appropriate.



3.(14) Consider the molecule 2(*R*),3(*S*) 2-bromo-3-chlorobutane which has a  $[\alpha]_D = -79^\circ$ .

a) Draw the Fisher projection of this molecule with  $C_2$  above  $C_3$ . Label it "**A**".

b) Draw the stereoisomer that has a  $[\alpha]_D = +79^\circ$ . Label it "**B**".

c) Draw 2 other stereoisomers and label them "**C**" and "**D**".

d) What is the stereochemical relationship between:

A & B \_\_\_\_\_

A & C \_\_\_\_\_

B & D \_\_\_\_\_

C & D \_\_\_\_\_

4.(18) Draw a staggered *Newman projection* of **2,3-dibromobutane** sighting along C<sub>2</sub>-C<sub>3</sub> where the 2 Br atoms are **anti** and the 2 methyl groups (C1 and C4) are **anti**.

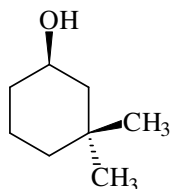
- a) Rotate the rear carbon in 60° increments and draw the other 5 conformations. Using the data below determine the relative energy of both conformations and the barrier to rotation (kJ/mol).

<u>Eclipsing</u>	<u>kJ/mol</u>	<u>Gauche</u>	<u>kJ/mol</u>
H – H	4.0	CH <sub>3</sub> – CH <sub>3</sub>	3.8
Br – H	5.0	CH <sub>3</sub> – Br	3.2
CH <sub>3</sub> – H	6.0	Br – Br	2.0
Br – Br	8.0		
CH <sub>3</sub> – Br	10.0		
CH <sub>3</sub> – CH <sub>3</sub>	11.0		

- b) How many chiral centers are in this compound? \_\_\_\_\_
- c) Determine the absolute configuration of any chiral centers.
- d) Is this particular molecule optically active? \_\_\_\_\_ Explain.

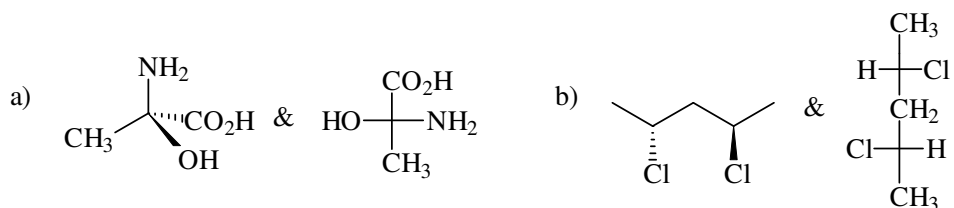
5.(16) The molecule 3,3-dimethylcyclohexanol is show below and consider the following information:

- i) CH<sub>3</sub> – H 1,3-diaxial interaction = 3.8 kJ/mol
- ii) OH – H 1,3-diaxial interaction = 2.1 kJ/mol
- iii) OH - CH<sub>3</sub> 1,3-diaxial interaction = 8.1 kJ/mol

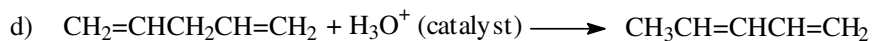
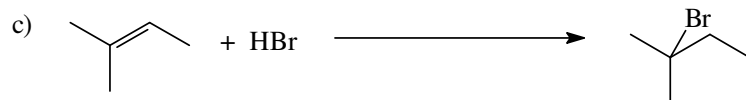
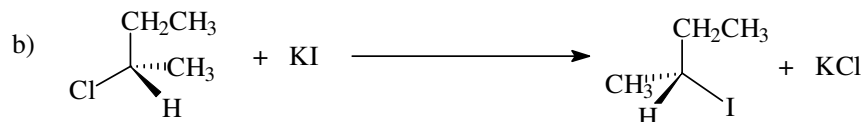


- a) Above, draw the two flip *chair conformations* and identify any interactions present in each. **Highlight any H's** that are involved in 1,3-diaxial interactions.
- b) Using the data provided, determine the relative energy of each conformation.

6.(14) Determine the absolute configuration (*R* or *S*) for each chiral center in the following pairs of molecules. Also determine if the pairs are *identical*, *enantiomers* or *diastereomers*.



7.(12) Identify the following chemical reactions as either *rearrangement*, *elimination*, *substitution* or *addition*. Also identify the electrophile (E) and the nucleophile (N) in each reaction.



8.(4) Briefly define the term “racemic mixture”. What is the specific rotation of a racemic mixture?